OXYGEN AND TEMPERATURE ELECTRODE



General features

The oxygen content in liquids is measured with a system called Clark's cells. These cells generate an electrical current proportional to the oxygen partial pressure which can be evaluated with a suitable measurement converter.

In order to prevent interference effects on measuring, the Clark's cells are covered with a gas-permeable membrane. The membranes typically used are made from PTFE but, as this material is mechanically fragile, frequent changing is often necessary, along with the related "demanding" operations (interruption of measurement, electrolyte replacement, regeneration of the electrodes).

The **\$423** solves this problem by using an OPTIFLOW™ membrane. This membrane is very mechanically stable, is manufactured as a laminate around a steel mesh and is very resistant to chemically aggressive environments as well as high pressures.

Thanks to the special construction of the measuring electrodes, this system also makes the sensor totally "maintenance free".

Applications

Surface waters, drinking water, biological treatment of waste water.

Technical specifications

Measuring range	40 ppb ÷ 40 ppm
Measuring method	measure of the electric current influenced by the oxygen partial pressure
Sensitivity	40 ÷ 80 nA a 25 °C in air
Stabilization time	typical 15 min., max. 1 h
Required flow rate	≥ 0.03 m/s
Temperature sensor	NTC 30 kOhm Oxysens W (NTC 22 kOhm Oxysens – optional)
Operating temperature	0 ÷ 60°C
Maximum pressure	0 ÷ 4 bar
Body material	Stainless Steel 1.4435, PEEK, Silicon, NBR
Electrode material	Silver-Platinum combination
Membrane material	OPTIFLOW
Reference electrolyte	Alkaline solution
Electrical connector	Integral cable 5 mt
Connection to process	Pg 13.5 threaded
Polarisation current	-670 +/- 50 mV

